

Internal Combustion Engines

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Exercise-Otto cycle

Below listed data are provided for a four stroke internal combustion engine which operates according to air-standard **OTTO** cycle. Please calculate the **pressure** and **temperature** of the characteristic cycle points and the indicated mean effective pressure (P_{mi}) of the cycle.

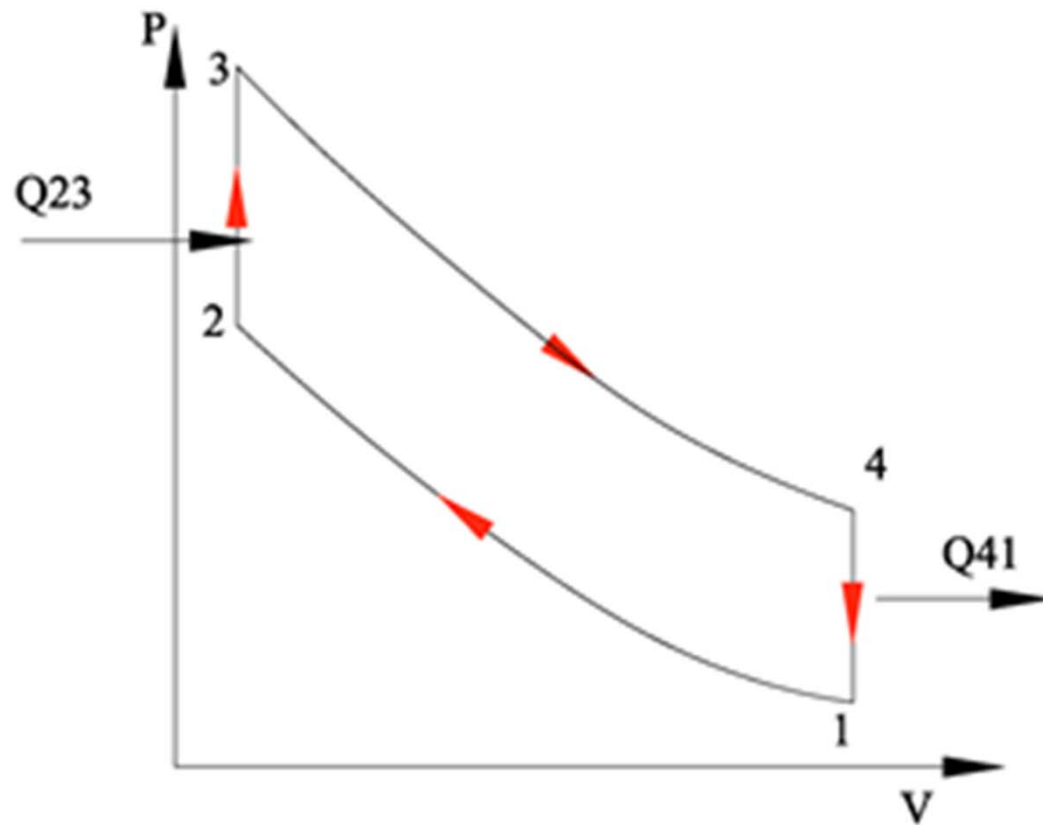
Given parameters,

- Temperature at the beginning of compression stroke= 20°C
- Pressure at the beginning of compression stroke= 1 Bar
- Compression Ratio is 10.5
- Pressure increase ratio is 2
- Ratio of specific heats is 1.41

(The SEILIEGER Cycle P_{mi} and η_T equations are given below. You can use these equations for calculating OTTO Cycle P_{mi} and η_T)

$$P_{mi} = \eta_T \cdot \frac{P_1 \cdot \epsilon^k}{(k-1) \cdot (\epsilon - 1)} \left[\rho - 1 + k \rho (\epsilon_g - 1) \right] \quad \eta_T = 1 - \frac{1}{\epsilon^{(k-1)}} \cdot \frac{\rho \cdot \epsilon_g^k - 1}{\rho - 1 + k \cdot \rho \cdot (\epsilon_g - 1)}$$

Exercise-Otto cycle



Exercise-Otto cycle

$$P_1 = 1 \text{ Bar}$$

$$P_2 = P_1 \cdot \epsilon^k$$

$$P_2 = 1 \cdot 10,5^{1,41} = 27,53 \text{ Bar}$$

$$\frac{P_3}{P_2} = \rho, \quad \frac{P_3}{27,53} = 2$$

$$P_3 = 55,06 \text{ Bar}$$

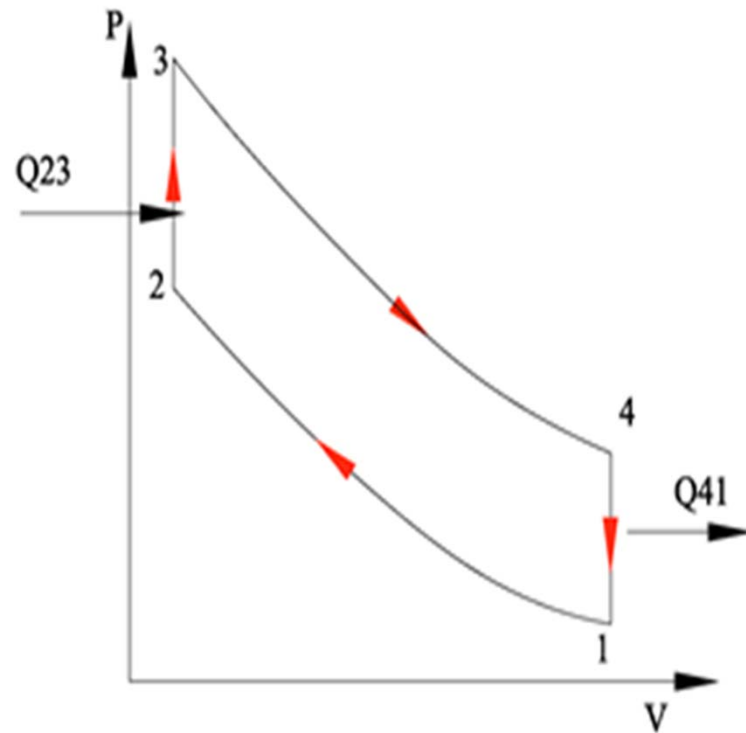
$$P_4 = \frac{P_3}{\epsilon^k}, \quad P_4 = \frac{55,06}{10,5^{1,41}}$$

$$P_4 = 2 \text{ Bar}$$

$$T_1 = 293 \text{ K}$$

$$T_2 = T_1 \cdot \epsilon^{k-1}$$

$$T_2 = 293 \cdot 10,5^{1,41-1}, \quad T_2 = 768,34 \text{ K}$$



Exercise-Otto cycle

$$T_2 = 293 \cdot 10,5^{1,41-1}, T_2 = 768,34 \text{ K}$$

$$\frac{T_3}{T_2} = \rho, \frac{T_3}{768,34} = 2$$

$$T_3 = 1536,68 \text{ K}$$

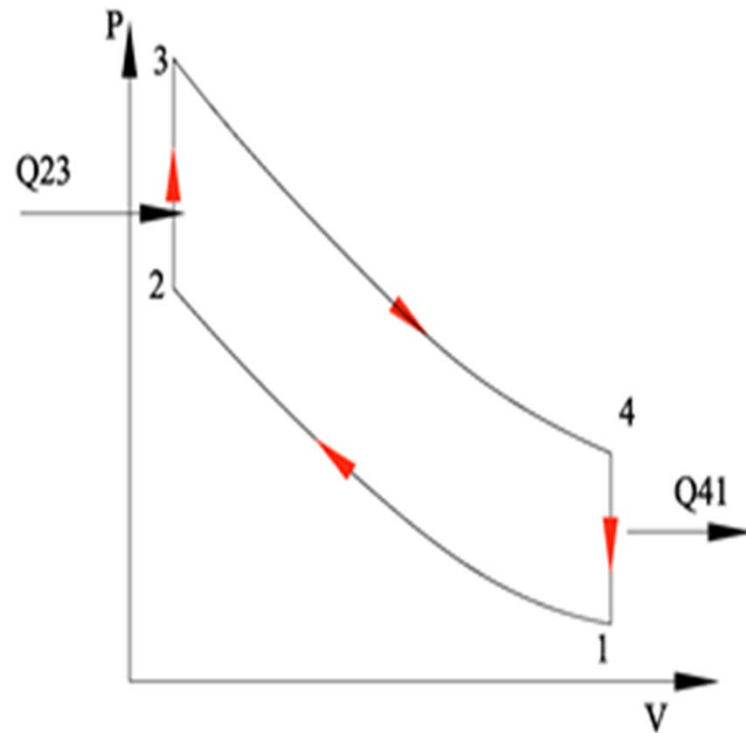
$$T_4 = \frac{T_3}{\varepsilon^{k-1}}, T_4 = \frac{1536,68}{10,5^{1,41-1}}$$

$$T_4 = 585,99 \text{ K}$$

$$\eta_T = 1 - \frac{1}{\varepsilon^{k-1}}, 1 - \frac{1}{10,5^{1,41-1}}, \eta_T = 0,6186$$

$$P_{mi} = \eta_T \cdot \frac{P_1}{k-1} \cdot \frac{\varepsilon^k}{\varepsilon-1} \cdot (\rho-1)$$

$$P_{mi} = 0,6186 \cdot \frac{1}{1,41-1} \cdot \frac{10,5^{1,41}}{10,5-1} \cdot (2-1) = 4,372 \text{ Bar}$$



Exercise-Diesel cycle

Below listed data are provided for a four stroke internal combustion engine which operates according to air-standard **DIESEL** cycle. Please calculate the **pressure** and **temperature** of the characteristic cycle points and the indicated mean effective pressure (P_{mi}) of the cycle.

Given parameters,

Pressure at the beginning of compression stroke is 1 Bar.

Temperature at the beginning of compression stroke is 27°C.

Cut-off Ratio is 1.4

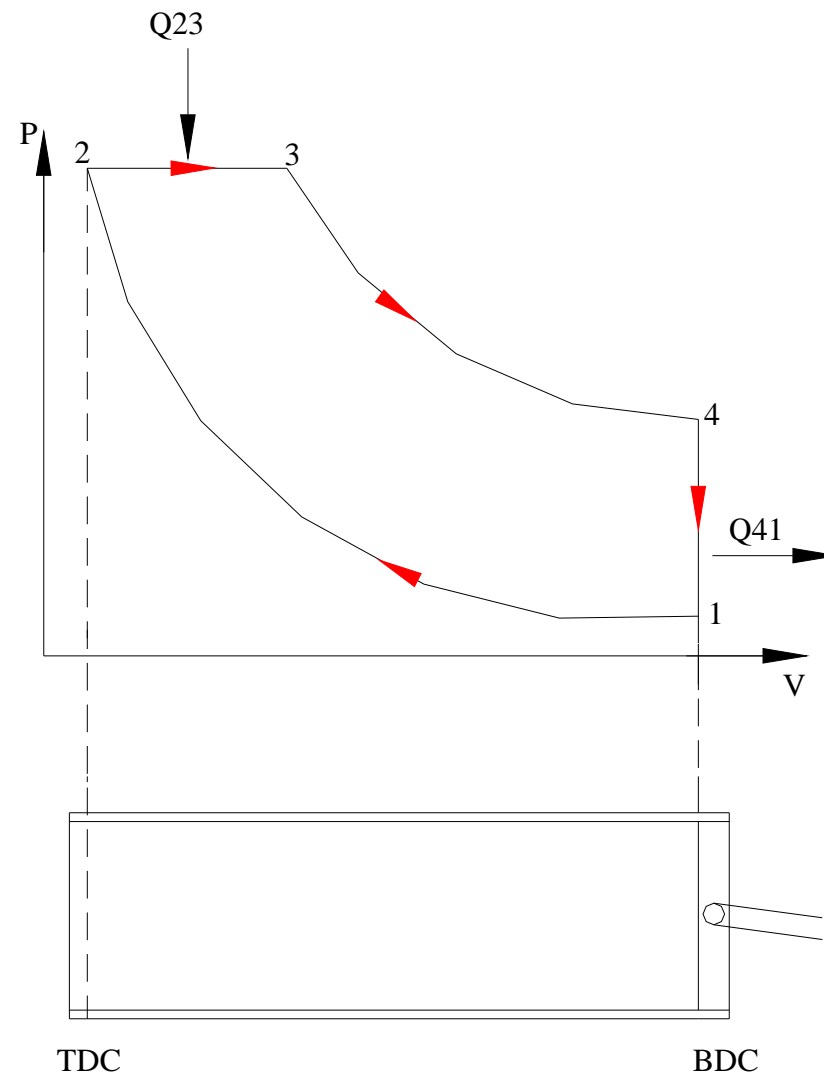
Ratio of specific heats 1.37

Pressure at the beginning of the heat input process is 57.3 Bars.

(The SEILIEGER Cycle P_{mi} and η_t equations are given below. You can use these equations for finding OTTO Cycle P_{mi} and η_t)

$$P_{mi} = \eta_T \cdot \frac{P_1 \cdot \epsilon^k}{(k-1) \cdot (\epsilon - 1)} \left[\rho - 1 + k \rho (\epsilon_g - 1) \right] \quad \eta_T = 1 - \frac{1}{\epsilon^{(k-1)}} \cdot \frac{\rho \cdot \epsilon_g^k - 1}{\rho - 1 + k \cdot \rho \cdot (\epsilon_g - 1)}$$

Exercise-Diesel cycle



Exercise-Diesel cycle

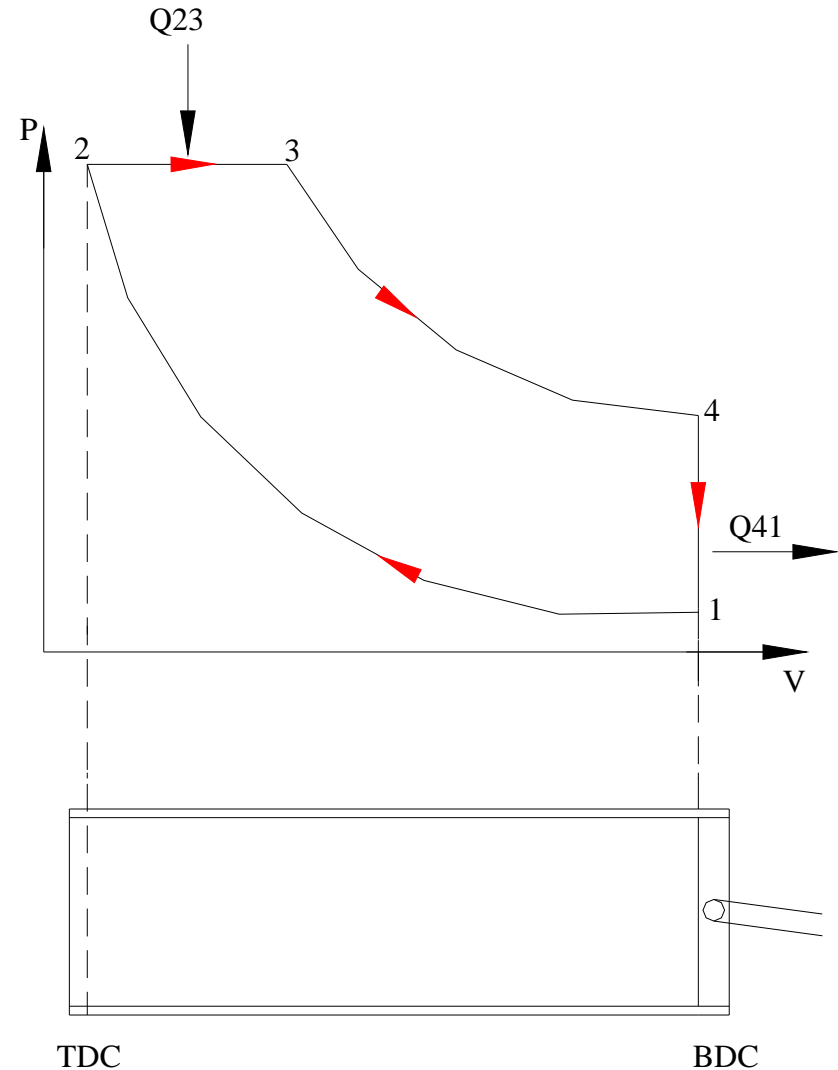
$$T_1 = 300K$$

$$P_1 = 1 \text{ Bar}$$

$$P_2 = 57.3 \text{ Bars}$$

$$\varepsilon_g = 1.4$$

$$k = 1.37$$



Exercise-Diesel cycle

$$P_2 = P_1 \cdot \varepsilon^k$$

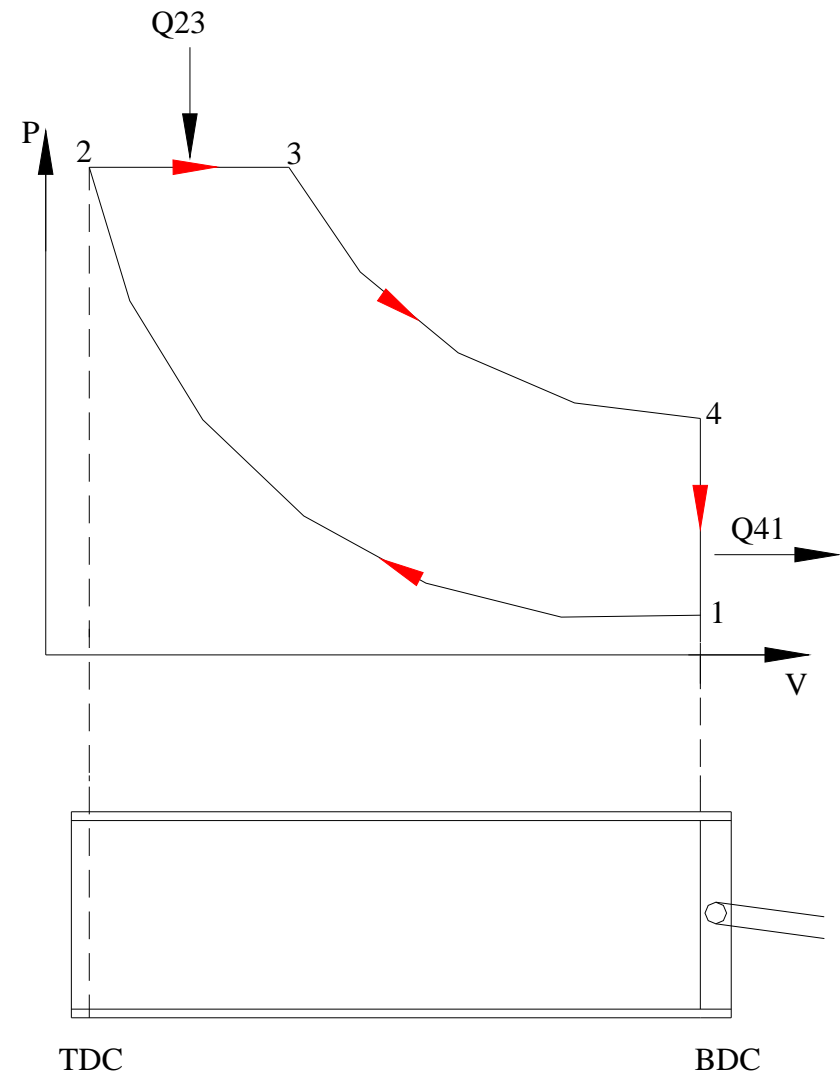
$$57.3 = 1 \cdot \varepsilon^{1.37}$$

$$\varepsilon = 19.2$$

$$P_2 = P_3$$

$$P_3 = 57.3 \text{ Bars}$$

$$P_4 = \frac{P_3}{\left(\frac{\varepsilon}{\varepsilon_g}\right)^k} = \frac{57.3}{\left(\frac{19.2}{1.4}\right)^{1.37}} = 1.57 \text{ Bars}$$



Exercise-Diesel cycle

$$T_2 = T_1 \cdot \varepsilon^{k-1}$$

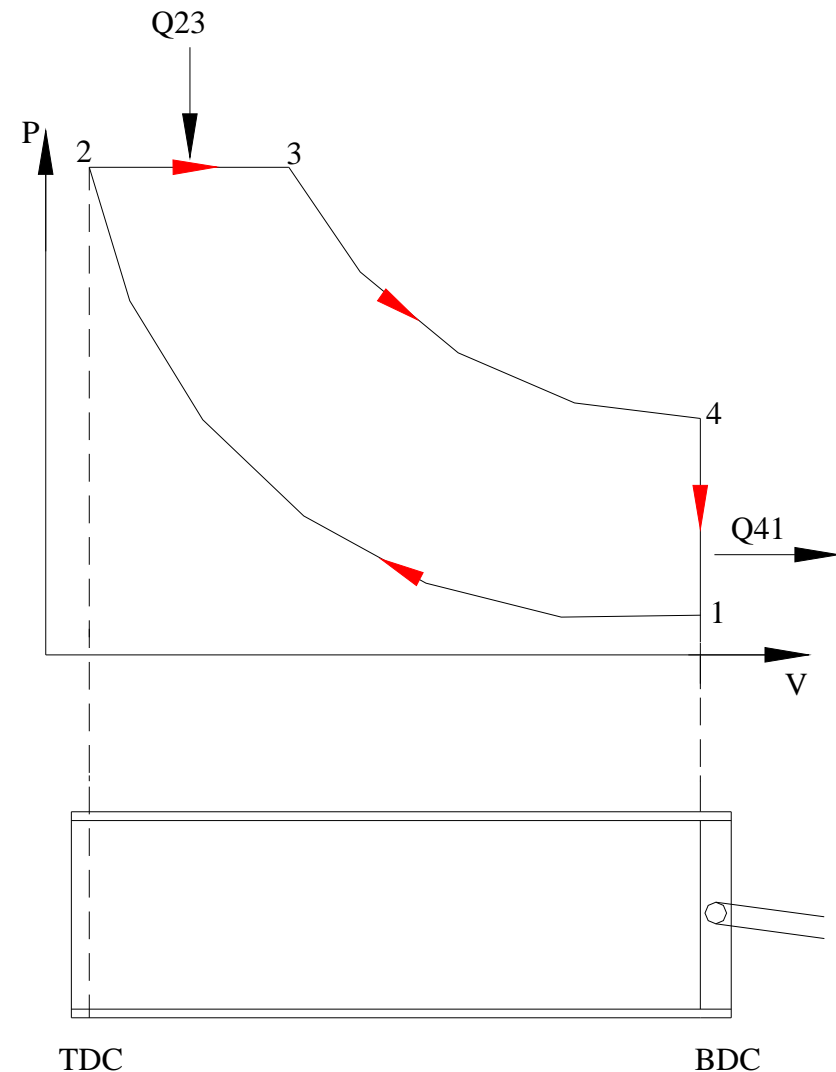
$$T_2 = 300.19.2^{1.37-1}$$

$$T_2 = 895.24K$$

$$T_3 = T_2 \cdot \varepsilon_g$$

$$T_3 = 895.24 \times 1.4$$

$$T_3 = 1253.3K$$

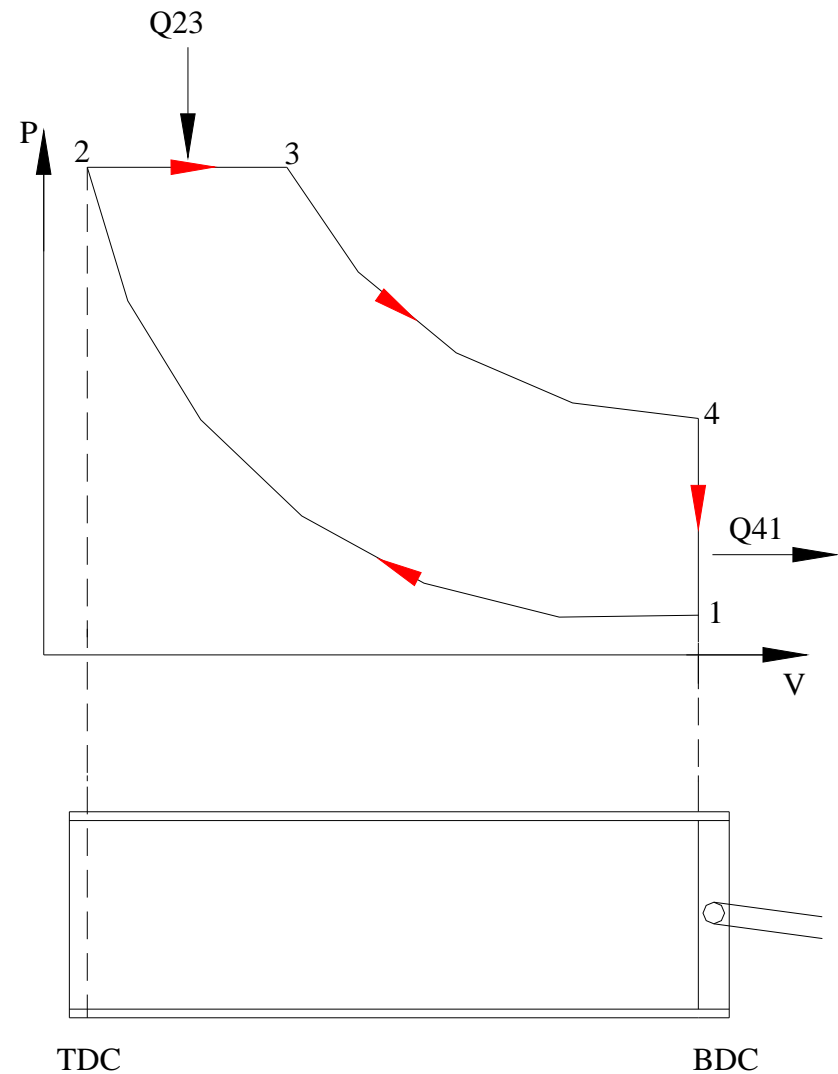


Exercise-Diesel cycle

$$T_4 = \frac{T_3}{\left(\frac{\varepsilon}{\varepsilon_g}\right)^{k-1}}$$

$$T_4 = \frac{1253.3}{\left(\frac{19.2}{1.4}\right)^{1.37-1}}$$

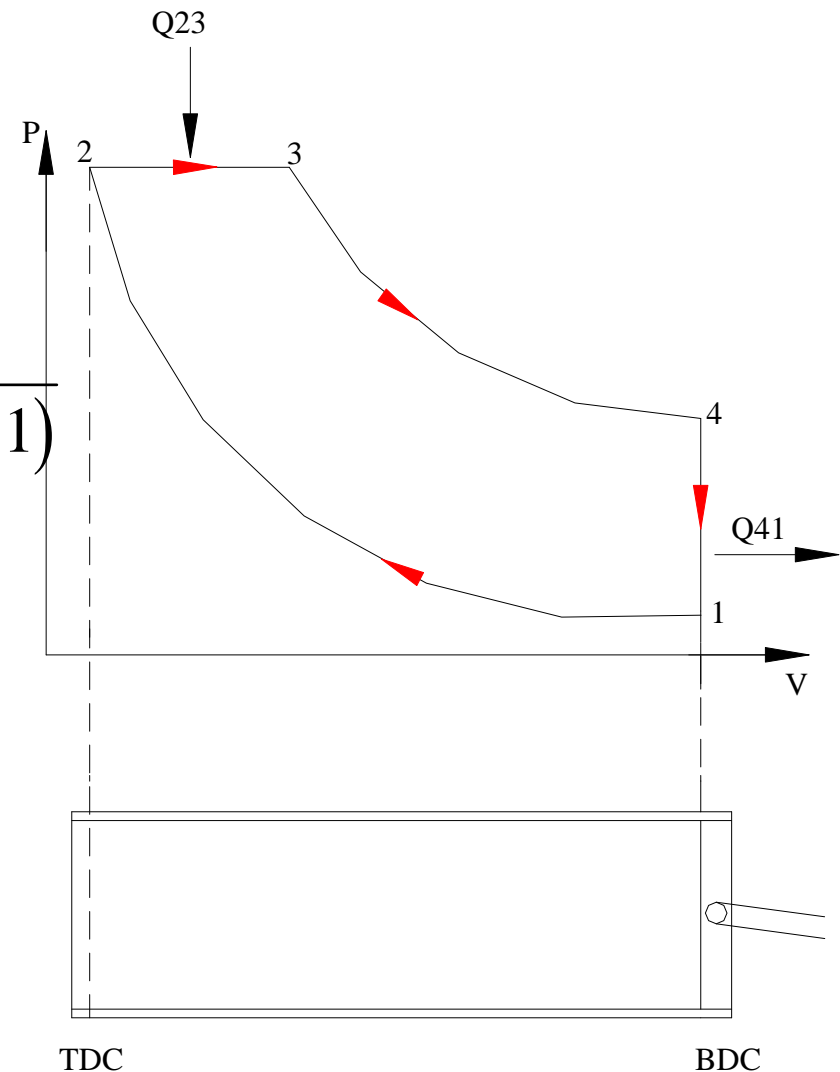
$$T_4 = 475.81K$$



Exercise-Diesel cycle

$$\eta_T = 1 - \frac{1}{\varepsilon^{(k-1)}} \cdot \frac{\rho \cdot \varepsilon_g^k - 1}{\rho - 1 + k \cdot \rho \cdot (\varepsilon_g - 1)}$$

$$\eta_T = 1 - \frac{1}{19.2^{(1.37-1)}} \times \frac{1 \times (1.4^{1.37}) - 1}{1 - 1 + (1.37) \times 1 \times (1.4 - 1)}$$



Exercise-Dual (Seiliger)cycle

Below listed data are provided for a four stroke internal combustion engine which operates according to air-standard **DUAL** cycle. Please calculate the **pressure** and **temperature** of the characteristic cycle points and the indicated mean effective pressure (P_{mi}) of the cycle.

Given parameters,

Pressure at the beginning of compression stroke is 1 Bar.

Pressure and temperature at the end of heat input process are 43.05×10^5 Pa and 2410K respectively.

Temperature at the end of expansion stroke is 1150K.

Compression ratio is 11.

Ratio of specific heats 1.41

(The SEILIEGER Cycle P_{mi} and η_T equations are given below. You can use these equations for finding OTTO Cycle P_{mi} and η_T)

$$P_{mi} = \eta_T \cdot \frac{P_1 \cdot \epsilon^k}{(k-1) \cdot (\epsilon - 1)} \left[\rho - 1 + k \rho (\epsilon_g - 1) \right] \quad \eta_T = 1 - \frac{1}{\epsilon^{(k-1)}} \cdot \frac{\rho \cdot \epsilon_g^k - 1}{\rho - 1 + k \cdot \rho \cdot (\epsilon_g - 1)}$$

Exercise-Dual (Seiliger)cycle

$$P_1 = 1.01 \cdot 10^5 \text{ Bars}$$

$$P_2 = P_1 \cdot \varepsilon^k$$

$$P_2 = 1.01 \cdot 10^5 \times 11^{1.41} = 29.6 \text{ Bars}$$

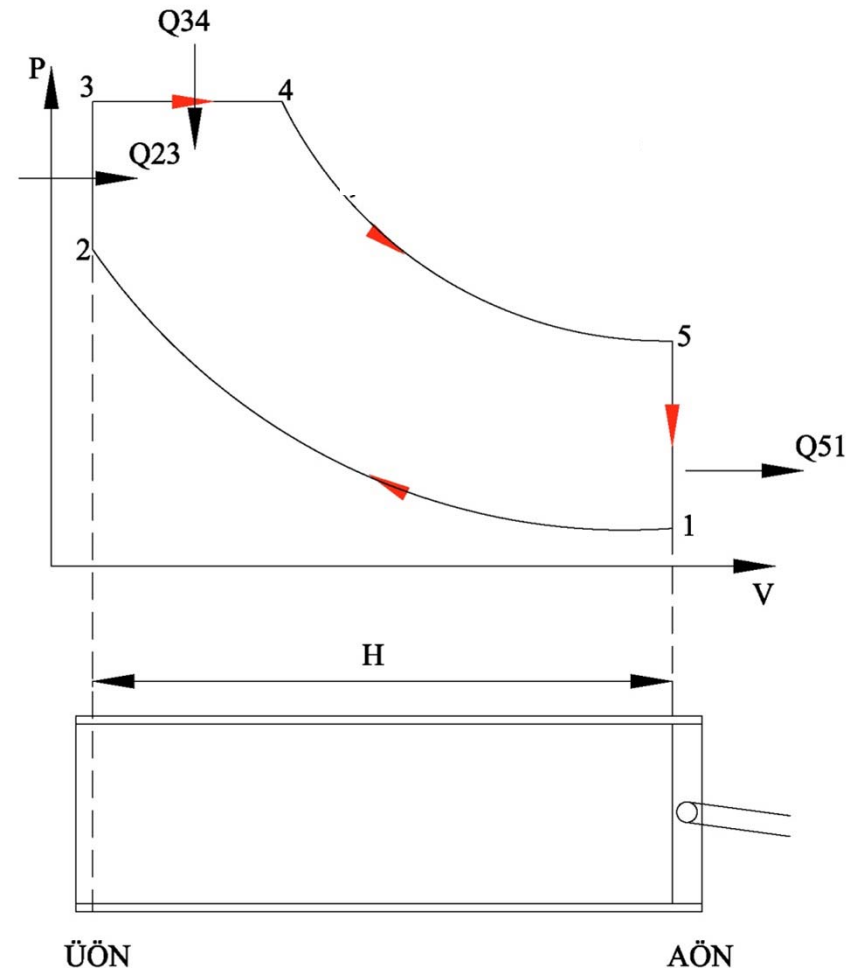
$$P_3 = P_4 = 43.05 \text{ Bars}$$

$$T_4 = 2410 \text{ K}$$

$$T_5 = 1150 \text{ K}$$

$$T_5 = \frac{T_4}{\left(\frac{\varepsilon}{\varepsilon_g}\right)^{k-1}} \quad 1150 = \frac{2410}{\left(\frac{11}{\varepsilon_g}\right)^{1.41-1}}$$

$$\varepsilon_g = 1,81$$



Exercise-Dual (Seiliger)cycle

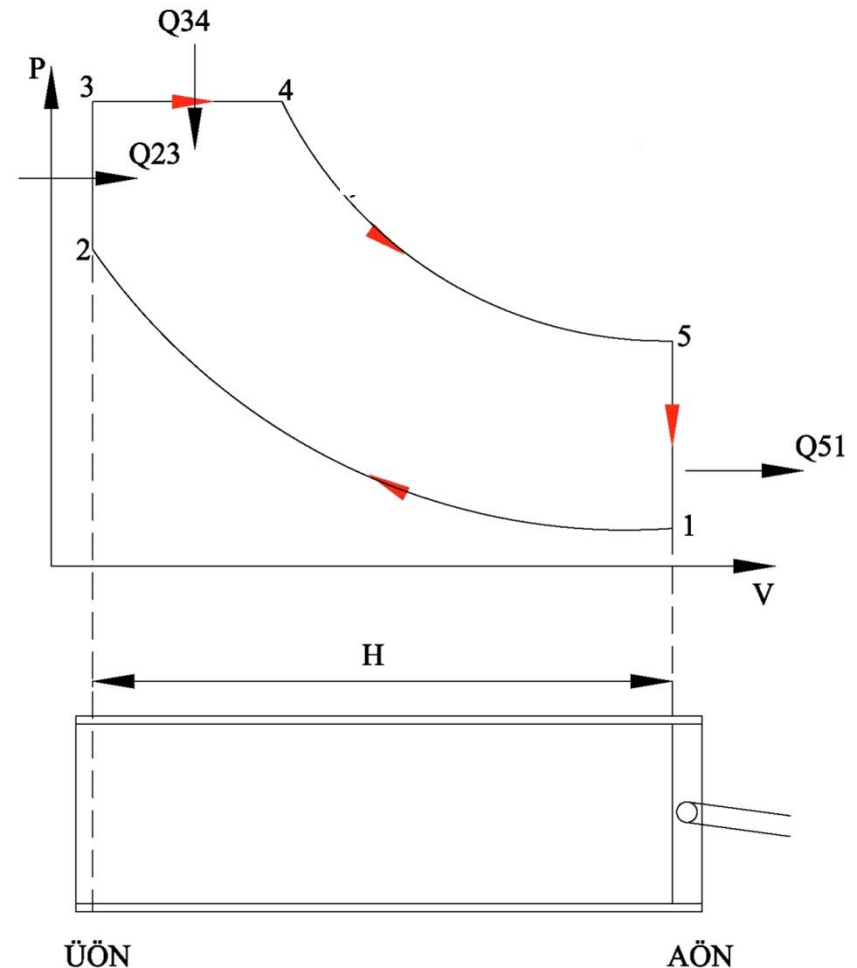
$$T_4 = T_3 \cdot \varepsilon_g$$

$$2410 = T_3 \cdot 1.81$$

$$T_3 = 1331.4K$$

$$\frac{P_3}{P_2} = \rho \quad \frac{43,5}{29,6} = \rho$$

$$\rho = 1,469$$



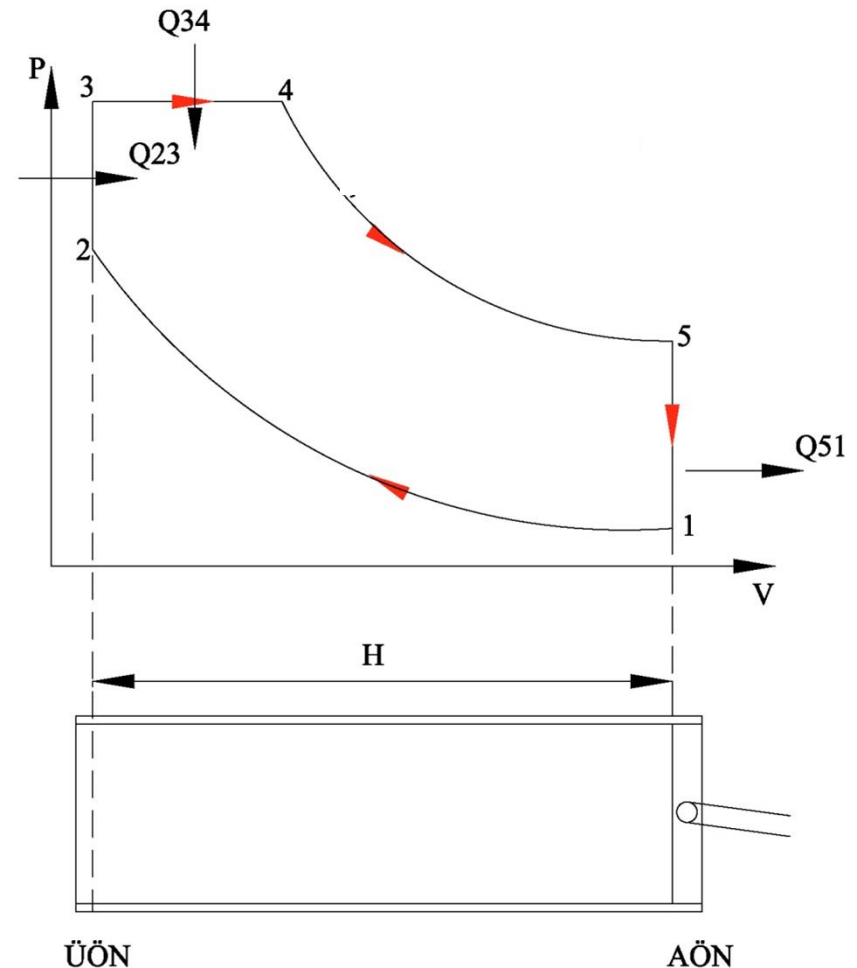
Exercise-Dual (Seiliger)cycle

$$\frac{T_3}{T_2} = \rho \quad \frac{1331.4}{T_2} = 1.469$$

$$T_2 = 906.39 K$$

$$T_2 = T_1 \cdot \varepsilon^{k-1} \quad 906,39 = T_1 \cdot 11^{0,41}$$

$$T_1 = 339,11 K$$

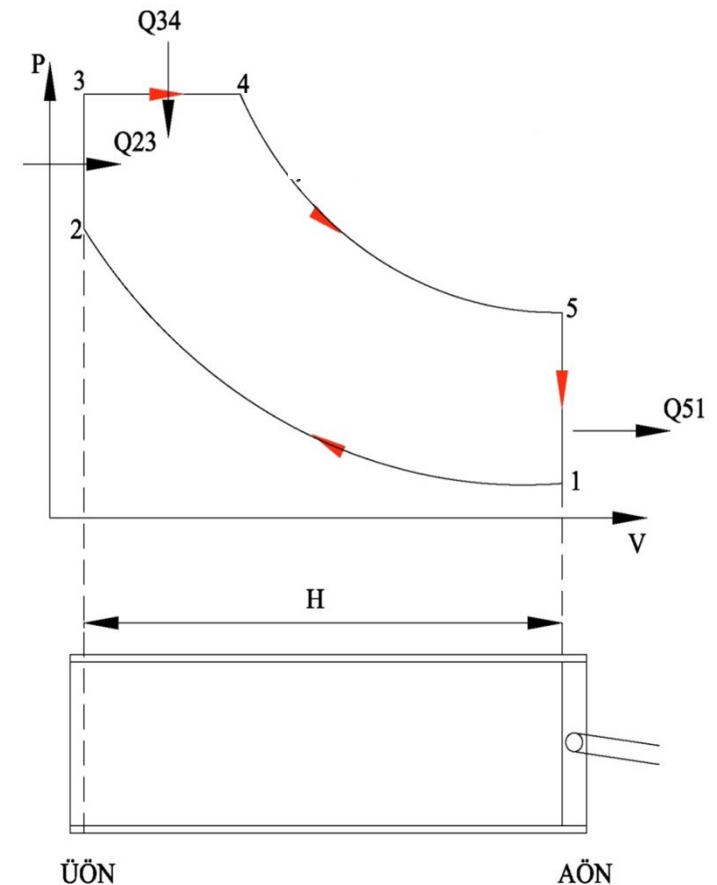


Exercise-Dual (Seiliger)cycle

$$\eta_T = 1 - \frac{1}{\varepsilon^{k-1}} \cdot \frac{\rho \cdot \varepsilon_g^k - 1}{\rho - 1 + k \cdot \rho \cdot (\varepsilon_g - 1)}$$

$$\eta_T = 1 - \frac{1}{11^{0,41}} \cdot \frac{1,469 \cdot 1,81^{1,41} - 1}{1,469 - 1 + 1,41 \cdot 1,469 \cdot (1,81 - 1)}$$

$$\eta_T = 0.5832$$



Exercise-Dual (Seiliger)cycle

$$Pmi = \eta_T \cdot \frac{P_1 \cdot \varepsilon^k}{(k-1) \cdot (\varepsilon-1)} [\rho - 1 + k\rho(\varepsilon_g - 1)]$$

$$Pmi = 0,5832 \cdot \frac{1,01 \cdot 11^{1,41}}{(1,41-1) \cdot (11-1)} [1,469 - 1 + 1,41 \cdot 1,469(1,81-1)]$$

$$Pmi = 9.06 \text{ Bars}$$

